

〈論文〉

Use of Exogenous Hydrogen for Thermophilic Digestion of Wastewater Sludge

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Abstract

The feasibility of using hydrogen (H_2) to upgrade biogas during the thermophilic digestion of wastewater sludge was investigated in batch experiments. A high H_2 consumption rate of 32.95~34.13 mL/($g_{VS} \cdot h$) was obtained at stirring speed 100 rpm. However, the addition of H_2 inhibited significantly the digestion process. Changing the setting of reactors from horizontal to vertical is an effective method for decreasing the rate of digestion inhibition from 93.70% to 54.24%. Moreover, the required recovery time for methane (CH_4) generation was much long, although digestion process could gradually recover once the hydrogen partial pressure (P_{H_2}) become very low. The inhibited digestion from exogenous H_2 was mainly attributed to the increased H_2 concentration in sludge rather than the increased pH or CO_2 depletion. Thus, adding H_2 in the headspace of anaerobic digester is a feasible method for upgrading biogas as well as avoiding the digestion inhibition.

Key words : hydrogen, digested sludge, methanogens, acid formation bacteria, syntrophic acetate-oxidizing bacteria

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1. Introduction

Biological methanogenesis from wastewater sludge under anaerobic conditions is a highly sustainable waste treatment process due to the conversion of organic content into renewable bioenergy in the form of methane (CH_4) before incineration or landfill treatment. Anaerobic digestion is a complex biological process typically involving hydrolysis, acidogenesis, acetogenesis, and methanogenesis¹⁾. Complex organic matter (primary polymers of carbohydrates and proteins) is first converted into soluble organic matter, and the hydrolysis products are subsequently fermented to form various intermediated products, such as volatile fatty acids (VFAs), followed by the conversion of these VFAs into acetic acid, hydrogen (H_2), and carbon dioxide (CO_2). Finally, methanogens utilize the acetic acid and H_2/CO_2 as substrates to produce CH_4 . **Fig. 1** shows the general pathway of the anaerobic digestion process¹⁾.

Although acetate is one of the methanogenic substrates, H_2 is also an important precursor of hydrogenotrophic methanogens, as it can be used as an electron donor for conversion of CH_4 by reducing CO_2 . During anaerobic digestion, about 30% of the CH_4 is

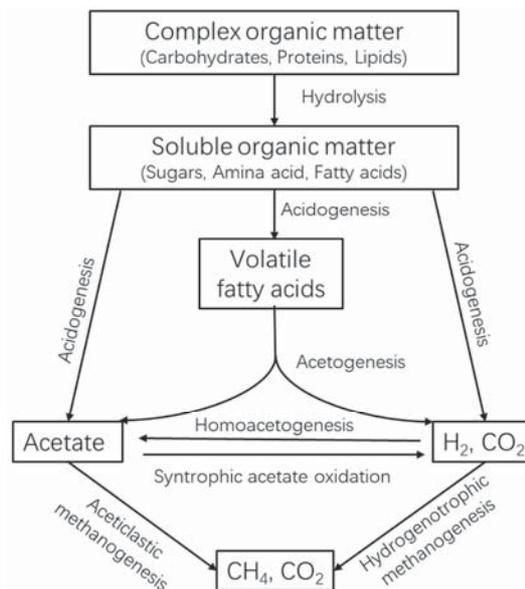
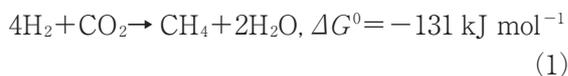


Fig. 1 General pathway of the anaerobic digestion process¹⁾

produced directly from H_2/CO_2 by hydrogenotrophic methanogens²⁾. Furthermore, some studies found that syntrophic acetate oxidation coupled to hydrogenotrophic methanogenesis was a dominant pathway in thermophilic methanogenic reactors³⁾. Thus, upgrading biological biogas (primarily CH_4 and CO_2) by adding exogenous H_2 can remove CO_2 and simulta-

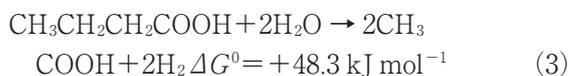
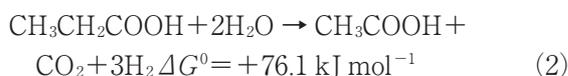
neously increase CH₄ yield by accelerating the activity of hydrogenotrophic methanogens.

H₂, an intermediate product of digestion, can be produced from organic waste via biological fermentation, but yields are limited⁴. Hydrogenotrophic methanogens are able to consume an equimolar amount of four times H₂ to CO₂ and generate CH₄ as per Eq. (1).



1 atm, 298 K, pH=7, H₂ partial pressure=1 atm,

However, anaerobic digestion is a complicated biochemical process catalyzed by a range of microorganisms. Therefore, although the addition of H₂ enhanced the biological conversion of CO₂ to CH₄ via hydrogenotrophic methanogens, the associated increase in dissolved H₂ would influence the production of H₂ in the metabolic pathways of bacteria. This is due to thermodynamics, which dictate the feasibility of fatty acid oxidation, such as from propionate (Eq. (2)) and butyrate (Eq. (3)) to acetate via acetogenesis⁵.



1 atm, 298 K, pH=7, H₂ partial pressure=1 atm, 1.0 mol/L of propionate and butyrate

Under standard conditions, these reactions remain endergonic and become feasible only when hydrogen partial pressure (P_{H₂}) is maintained below 10 Pa⁵. This extremely low P_{H₂} was sustained by interspecies H₂ transfer between acetogens and hydrogenotrophic methanogens. This balance of immediate production and consumption of H₂, and the associated transfer of electrons, is affected by the addition of exogeneous H₂.

Several studies have applied H₂ with the aim of improving biogas quality⁴. One approach is *ex-situ* upgrading in an adjacent external reactor, and another involves injection of H₂ into the reactor (*in-situ* upgrading). In an *ex-situ* system, the initial stages of anaerobic digestion (hydrolysis and acidogenesis) are not present. Thus, reactor stability and performance depend only on sufficient provision of CO₂ (e. g., via a fermentation process), H₂, essential nutrients, and hydrogenotrophic methanogens. However, in an *in-*

situ digestion system, H₂ would reduce the biochemical potential of anaerobic digestion. Successful upgrading of biogas processes *in situ* by the addition of H₂ during anaerobic digestion have been reported. Luo's study showed that, in the case of thermophilic co-digestion of manure and acidic whey, the pH of the anaerobic reactor could be maintained below 8.0 with the addition of H₂, which did not inhibit the anaerobic process⁶. In Agneessens's study, when the headspace CO₂ concentration increased above 12%, CH₄ production recovered in a mesophilic co-digestion of sludge and straw, although the pH was still 8.27 ± 0.05. This suggests that the inhibition of hydrogenotrophic methanogens due to a low CO₂ concentration was more pronounced than the effect of the increase in reactor pH⁷. Alfaros's study showed that organic matter removal in a mesophilic upgrading reactor for thickened mixed sludge was not compromised by supplying H₂, or by the high pH level when H₂ was added through a submerged membrane that make H₂ utilized by the biofilm attached on the surface of the membranes⁸.

Wastewater sludge, as a by-product of wastewater treatment, can be recovered as energy. There are two methods of recycling wastewater sludge as fuel: carbonization (biochar) and digestion (biogas)⁹. Biochar can be sold to coal fire power plants, but it is very cheap and wastewater treatment plants (WWTPs) sometimes have to pay the transportation costs. To compensate, CH₄ in biogas could supply electricity to WWTPs or even the district grid. Upgrading biogas by the addition of H₂ (e. g., via electrolysis of surplus renewable electricity) would promote energy recovery from sludge and thus indirectly reduce greenhouse gas emissions.

In our pervious study¹⁰, the effect of H₂ addition on mesophilic digestion of wastewater sludge was investigated in batch experiments. For the mesophilic digestion of wastewater sludge, addition of H₂ enhanced the conversion of CO₂ to CH₄; H₂ addition increased pH value and soluble chemical oxygen demand (SCOD); H₂ addition did not influence acetic acid degradation, but inhibited butyric acid degradation which is a substrate of acetogenesis; CH₄ production decreased as the H₂ injection frequency increased if the amount of injection is constant. Inhibition of H₂ can be alleviated if H₂ was added in the later period of a 24-h digestion experiment in that moment acetogenesis is almost complete. But CH₄

production from exogenous H₂ was limited since H₂ consumption rate is low.

Thermophilic digestion has more efficient biogas production capacity compared to mesophilic. The higher heat allows the digestion to operate at greater efficiency, handling more organic material in less time as the rate of bacterial activity increases with temperature. This increased efficiency can reduce digester retention times. But on the other side, thermophilic digestion is more sensitive since its biodiversity of the microbiome is lower. Consequently, the performance of thermophilic digestion with H₂ addition may differ from that of mesophilic digestion.

In this study, batch experiments were carried out to evaluate the H₂ consumption rate and effect of exogenous H₂ on the digestion process. The aim of this work was to determine whether the biogas produced by the thermophilic digestion of wastewater sludge could be upgraded by the addition of H₂.

2. Materials and methods

2.1 Sludge samples

Thermophilic digested sludge collected from a thermophilic digestion reactor in a WWTP (WWTP A) in Osaka City was used as a source of inoculum. The characteristics of the digested sludge are summarized in **Table 1**.

Table 1 Characteristics of thermophilic digested sludge

	Thermophilic digested sludge
TS (g/L)	17.93~18.10
VS (g/L)	12.17~12.25
pH	8.09~8.14
SCOD (mg/L)	2530~2580
Acetate (mg/L)	39.19~40.20
C (%)	31.45~31.46
H (%)	5.27~5.36
N (%)	4.86~4.97

2.2 Batch experiments

The preliminary experiment aimed to determine the CH₄ potential of sludge and optimum time scales for the experiments; The procedure is the same as the previous study¹⁰: 122-mL serum bottles were used as reactors and 42-mL thermophilic digested sludge was added to each bottle as an inoculum. The bottles were purged with N₂ (99.5%; Sumitoto Seika Chemicals) for 2 min and then sealed with butyl rubber stoppers and aluminum crimps. Three groups of reactors were



Chart 1 Serum bottle was were placed horizontally

placed horizontally (**Chart 1**) in the rotary shaker (BT300 water bath incubator; Yamato) at 55 °C and shaken at 100 rpm.

Batch experiment 1 aimed to estimate the H₂ consumption rate under thermophilic conditions. The reactors with sludge were prepared in the same way as in the preliminary experiment. The bottles were then incubated at 55 °C for 1 h to stimulate the microorganisms in the sludge. Then, 80 mL H₂ (99.99%; GL Science) was injected into the bottle using a syringe. The bottles were placed horizontally in the rotary shaker at 55 °C and shaken at two different speeds (50 and 100 rpm).

Batch experiment 2 was carried out to investigate the effect of the addition of H₂ on VFA degradation. The reactors with sludge were prepared in the same way as in the preliminary experiment. The bottles were incubated at 55 °C for 24 h to decompose the original VFA into sludge. Then, different volumes of H₂ were injected into the bottles. Three conditions were tested (A: 1 mL of water added to the sludge, B: 1 mL of 0.446 mol/L acetic acid added to the sludge, C: 1 mL of 0.223 mol/L butyric acid added to the sludge). Acetic acid and butyric acid were used in experiment 2 since they are substrate of methanogenesis and acetogenesis, respectively. And the does of acid is excessive. These liquids and H₂ (0, 20, 40, 60, 80 mL) were injected using a syringe. The three groups of reactors were placed horizontally in the rotary shaker at 55 °C and shaken at 100 rpm.

Batch experiment 3 was carried out to determine how to relieve digestion inhibition due to exogenous H₂; 122 mL serum bottles were used as reactors and 42 mL thermophilic digested sludge was added to each bottle. After that, the bottles were divided into four groups (D, E, F and G). Propionic acid was used to adjust the pH in group D to 6.55, 7.04, and 7.47. In group E, different volumes (10, 15, 20, and 30 mL) of CO₂ were injected into the bottles. The bottles were purged with N₂ for 2 min and then sealed with butyl rubber stoppers and aluminum crimps, and the bottles were incubated at 55 °C for 1 h. Then, 60 mL H₂ was injected into the bottles in groups D and E, which were

placed horizontally in the shaker at 55 °C (shaking speed=100 rpm). In group F, 60 mL H₂ was injected into the bottles at the beginning of the experiment, 3, 6, and 12 h intervals, respectively. The bottles were also placed horizontally in the shaker at 55 °C (shaking speed=100 rpm). In group G, different volumes (40, 60, 80, and 100 mL) of H₂ were injected into the bottles, which were placed vertically in the rotary shaker 100 rpm.

Three experiments were performed in triplicate; bottles containing only digested sludge without H₂ were used as controls.

2.3 Analytical procedures

The soluble chemical oxygen demand (SCOD), total solids (TS) and volatile solids (VS) were determined according to the Standard Methods.¹¹⁾ Gas samples were taken from the bottles to analyze the CO₂, H₂ and CH₄ contents by micro gas chromatography (micro GC) (490 Micro GC; Agilent Technologies). The pH was measured using a pH meter (D-71S; Horiba). VFAs were measured using high-performance liquid chromatography (HPLC; CDD-10AVP; Shimadzu). Prior to the analysis, the samples were centrifuged at 8,000 rpm for 10 min, and the supernatant was then filtered using a 13-mm-diameter filter (pore size=0.45 μm) to remove any remaining contaminants. CHN analysis was performed using an elemental analyzer (Micro Corder JM10; J-Science Lab).

3. Results and discussion

In this study, biogas volume is the total gas volume measured by syringe of appropriate range (**Chart 2**).



Chart 2 The gas volume was measured by syringe

The volume of each gas was calculated based on total gas volume and the concentration analyzed by micro GC.

Fig. 2 shows the biogas production of 42 mL sludge at 8, 24, 48, 72 h. The CH₄ production of the digested sludge was almost complete after 24 h and degradable organic compounds were scarce in the samples.

The amount of H₂ in the following H₂ addition experiments was based on the CO₂ yield. The biogas (primarily CH₄ and CO₂) yield of the digestion reactor in WWTP A plant is 0.59~0.73 Nm³/m³_{sludge}, and the average CO₂ content is 39.2%.¹²⁾ **Table 2** shows the H₂ requirements if all CO₂ is covered with H₂. Under the basis of these values, between 40 and 80 mL of H₂ was added for a sludge volume of 42 mL.

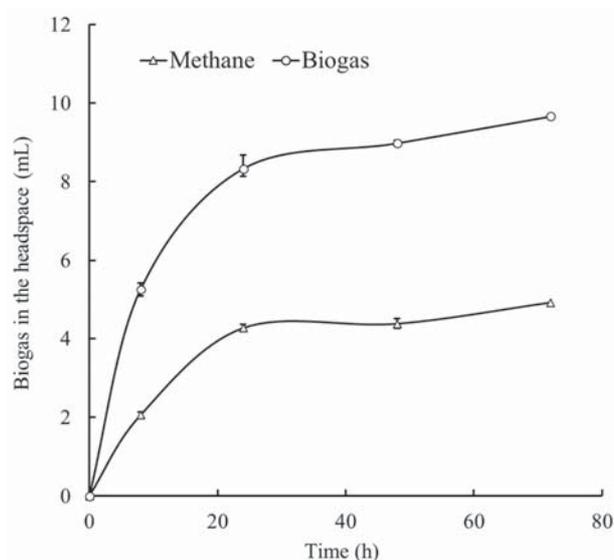


Fig. 2 Biogas productions of 42 mL sludge at 8, 24, 48, and 72 h

Table 2 Requirements for H₂ to react with all of the CO₂

	CO ₂ yield (mL/mL _{sludge} /d)	H ₂ requirement (mL/mL _{sludge} /d) ^a	H ₂ requirement (mL/mL _{sludge} /d) ^b
Min	0.23	0.92	1.56
Max	0.29	1.16	1.80

a: CH₄ from H₂ consumed based on the Sabatier equation (4 mol of H₂ is required to convert 1 mol of CO₂ into CH₄)

b: The amount of dissolved CO₂ in sludge is 0.16 mL/mL_{sludge}, based on the mole fraction of CO₂ in the liquid phase of 0.00131, given that the partial pressure of CO₂ is 40 kPa when the temperature is 55 °C¹¹⁾

3.1 Hydrogen consumption rate

In experiment 1, 80 mL of H₂ was added to the sludge and digested at different shaking speeds. H₂ uptake was rapid, with 79.83% and 100% of H₂ being consumed within 5 h at 50 rpm and 100 rpm, respectively. The amount of dissolved H₂ can be ignored because the solubility of H₂ in water is very low (18.2

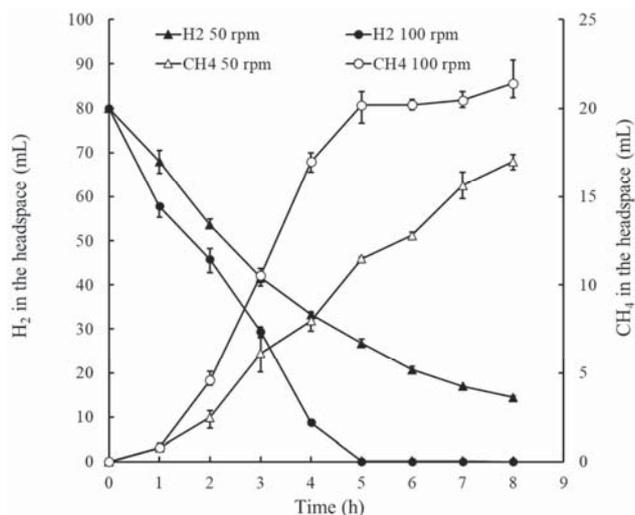


Fig. 3 Variations of H₂ and CH₄ in the headspace with digestion time in batch experiments carried out at 55 °C, at shaking speeds of 50 and 100 rpm

mL/L, 20 °C, 1 atm). **Fig. 3** shows that increasing the shaking speed to improve the H₂ gas-liquid mass transfer rate could accelerate the rate of conversion from H₂ to CH₄. The H₂ consumption rate (32.95~34.13 mL/(g_{VS} · h) was evaluated at 100 rpm and a linear downward trend was seen, with microbial activity or CO₂ production rate being the limiting factor, not the H₂ gas-liquid mass transfer rate¹⁴. Compared to our previous results for mesophilic digestion, in which the H₂ consumption rate was only 9.42~11.22 mL/(g_{VS} · h), thermophilic digestion had a greater ability to upgrade biogas through biological conversion of CO₂ to CH₄ via the addition of H₂.

According to Eq. (1) 4H₂+CO₂=CH₄+2H₂O, the limiting factor is microbial activity or CO₂ production rate when H₂ is sufficient. Given that a shortage of CO₂ may limit the consumption rate of H₂, another experiment was conducted after 24 h preliminary digestion to produce more CO₂, but a similar H₂ consumption rate was obtained. The result showed that the gas production didn't increase with CO₂ increase. It can indicate that limiting factor of H₂ consumption rate is hydrogenotrophic methanogenic activity when H₂ is sufficient.

3.2 Effects of exogenous hydrogen on digestion

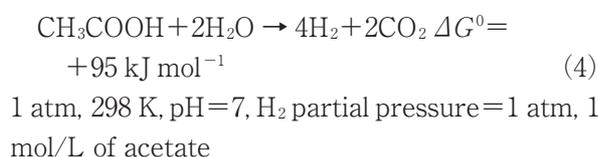
Two processes can be applied to utilize acetate in anaerobic digestion. The first process is acetoclastic methanogenesis: in this reaction, acetate is cleaved to methyl and carboxyl groups. The methyl group is directly converted to CH₄ via several biochemical reactions, whereas the carboxyl group is oxidized to

CO₂¹⁴. This reaction is not affected by the addition of H₂, because H₂ is neither a product nor a substrate.

Homoacetogens are another group of anaerobes that can use H₂ and CO₂ to form acetate instead of CH₄. When exogenous H₂ is introduced into an anaerobic digestion process, the elevated P_{H₂} stimulates homoacetogenesis¹⁵.

In batch experiment 2, no H₂ was observed in group A after 8 h, and little change in acetic acid concentration was seen: Acetic acid concentration was 42.1, 43.0, 41.2, 41.4, 40.3 mg/L when addition of H₂ was 0, 20, 40, 60, 80 mL, respectively. and; this indicated that there is no acetic acid accumulation, and the added H₂ had been completely consumed and converted into CH₄. Even if H₂ was consumed by homoacetogenesis, it contributed to acetate production and subsequent acetoclastic CH₄ formation.

The second process is syntrophic acetate oxidation¹⁶. In this reaction, acetate is converted to H₂ and CO₂ by syntrophic acetate-oxidizing bacteria, as shown in Eq. (4). This reaction requires close cooperation with hydrogenotrophic methanogens. The direct consumption of H₂ by hydrogenotrophic methanogens drives the thermodynamically unfavorable acetate oxidation:



The increasing H₂ concentration caused by exogenous H₂ inhibits the activity of syntrophic acetate-oxidizing bacteria, resulting in the inhibition of acetic acid degradation. However, the inhibition of exogenous H₂ on syntrophic acetate-oxidizing bacteria could be eliminated upon depletion of H₂ (H₂ was consumed by hydrogenotrophic methanogens).

The results of batch experiment 2 for group B revealed (**Fig. 4**) CH₄ production over the 8-h period following the addition of H₂ and acetic acid after 24 h of preliminary digestion. After 24 h of preliminary digestion, CH₄ production from sludge is rare, so most of the CH₄ produced after 24 h came arose the exogenous H₂ and acetic acid. The CH₄ produced from acetic acid can be calculated by subtracting the CH₄ produced from exogenous H₂ from the total amount of CH₄ produced. No H₂ was observed, indicating that the added acetic acid was utilized during acetoclastic methanogenesis

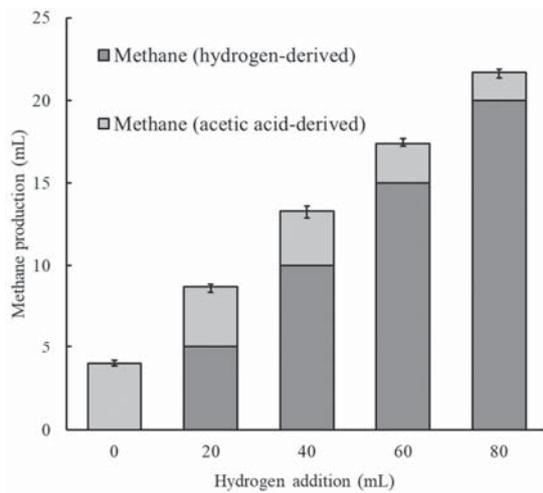


Fig. 4 CH₄ production within an 8-h period with different amounts of added H₂ added in group B (sludge + acetic acid)

or syntrophic acetate oxidation combined with hydrogenotrophic methanogenesis. The production of CH₄ from acetic acid can indicate the degree of degradation of acetic acid.

During an 8-h period, 3.99 mL CH₄ was converted from acetic acid without the addition of H₂, while CH₄ produced from acetic acid decreased to 93.01%, 82.47%, 60.17%, 42.91% when 20, 40, 60, 80 mL H₂ was added, respectively. This indicated that the addition of H₂ could indirectly affect hydrogenotrophic methanogenesis by inhibiting syntrophic acetate oxidation.

We observed CH₄ production in group C after 24 h because butyric acid is more difficult to degrade than acetic acid. After the 24-h preliminary digestion period, most of the CH₄ produced arose from the exogenous H₂ and butyric acid, for the same reason as in group B. Acetic acid converted from butyric acid can be neglected, because it can be utilized to form CH₄ via acetoclastic methanogenesis. The amount of CH₄ produced from butyric acid can be calculated by subtracting the CH₄ produced from exogenous H₂ from the total amount of CH₄ produced. However, according to **Fig. 5**, when 20 mL of H₂ was added, 58.2% more butyric acid was converted into CH₄ compared to the reactor without added H₂. This shows that exogenous H₂ strongly inhibited the degradation of butyric acid, suggesting that H₂ inhibited the acetogenesis process. Adding H₂ to a biogas reactor may cause problems due to an increase in the P_{H₂} of the biogas reactor, which inhibits acidogenesis.

When more than 40 mL H₂ was injected, the degradation of butyric acid was completely inhibited. This could be attributed to the higher P_{H₂}, higher pH (8.09, 8.41 and 8.59 with the addition 40 mL, 60 mL

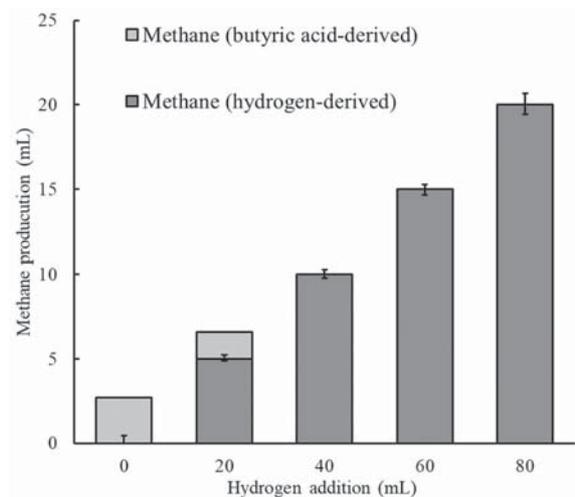


Fig. 5 CH₄ production within an 24-h period with different amounts of H₂ added in group C (sludge + butyric acid)

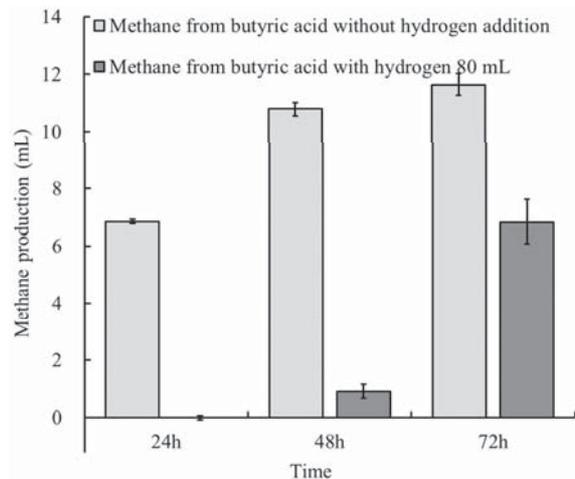


Fig. 6 CH₄ production within a 72-h period with the addition of H₂ and butyric acid in group C (sludge + butyric acid)

and 80 mL H₂, respectively), or depletion of CO₂, which rely on CO₂ as a carbon source.

The experimental time was extended to investigate the recovery of digestion following the addition of H₂ (**Fig. 6**); 80 mL H₂ was consumed by 42 mL sludge within the first 5 h, but butyric degradation took a long time to recover. Butyric acid started to degrade into CH₄ by the 48th h. During the following 24 h, CH₄ production in group C returned to normal compared to the first 24 h without added H₂ (**Fig. 6**).

To test the hypothesis that low concentrations of CO₂ could inhibit hydrogenotrophic methanogen substrates, in batch experiment 3 the pH was adjusted and CO₂ was added.

3.3 Cause of the digestion inhibition induced by hydrogen addition

The aim of this research was to upgrade biogas in an

anaerobic digester and minimize the inhibition of digestion. There are three possibilities regarding the effect of the addition of H₂ on digestion.

- (1) H₂ injection increases pH (because of CO₂ removal) which may limit the methanogenic activity ;
- (2) Depletion of CO₂ could inhibit the substrate for hydrogenotrophic methanogens, which relies on CO₂ as a carbon source ;
- (3) An increase in dissolved H₂ can subsequently inhibit VFA degradation.

These phenomena were verified in subsequent experiments.

The consumption of H₂ and CO₂ by hydrogenotrophic methanogens could increase the pH. According to Luo, in the case of co-digestion of manure with acidic whey, the pH in the anaerobic reactor with H₂ addition could be maintained below 8.0, which did not inhibit the anaerobic process⁶⁾.

Propionic acid was used to adjust the pH in this experiment, because it is more difficult to degrade than acetic and butyrate acid. Different amounts of propionic acid were added to the sludge, and 80 mL H₂ was then injected into the bottles. Bottles containing only sludge and propionic acid, without H₂, were used as controls. **Fig. 7** shows that reducing the initial pH in group D did not relieve the inhibition of CH₄ production.

The pH value could be maintained by supplying CO₂, which also stimulated hydrogenotrophic methanogens to consume H₂ and, subsequently, syntrophic acetate oxidation. However, we observed the same results in group E as in group D, namely that the inhibition was not suppressed.

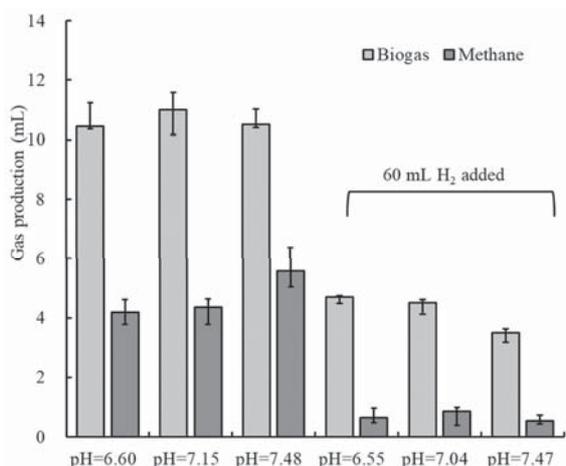


Fig. 7 Group D : biogas production from sludge with pH adjust (initial) after a 24-h digestion period

Under the premise that the dissolved H₂ concentration has an impact on sludge, 122-mL serum bottles (diameter : 40 mm, height : 130 mm), used as reactors, were placed horizontally in the rotary shaker and shaken at 100 rpm to consume H₂ within a short time (80 mL H₂ was consumed by 42 mL sludge in 5 h), because a large gas-liquid interface ensures sufficient mixing of H₂ and sludge. This operation reduces the H₂ consumption time, but all microorganisms, especially VFA-degrading bacteria, are exposed to high concentrations of H₂. Group F was carried out to was performed to determine the effect of P_{H2} on CH₄ production. H₂ was supplied frequently to lower H₂ partial pressure, but it had no effect on gas production even if P_{H2} went down by 1/8 (injection every 3 h). This result was different from that in mesophilic digestion condition in which CH₄ production decreased as the H₂ injection frequency increased if the amount of injection is constant. The possible reason is that difference of the microbial community in mesophilic and thermophilic digestion system.

On the other hand, if only part of the sludge is exposed to H₂, the H₂ can be consumed (80 mL H₂ was consumed by 42 mL sludge in 24 h) such that the degradation of organic matter may proceed without interference. When the serum bottles were placed vertically in the rotary shaker, the small gas-liquid interface allowed only the sludge on the surface to be exposed to H₂ due to the shape of the serum bottles. **Table 3** shows that in group G, CH₄ production from sludge was increased when all the bottles were placed vertically in the rotary shaker and shaken at 100 rpm comparing to the results in Group A which were shaken at the same speed but placed horizontally.

The inhibition rate of digestion has been used in our previous study to describe the performance of digestion with H₂ addition¹⁰⁾.

Theoretically, one-unit volume of CO₂ could convert

Table 3 CH₄ and CO₂ production with different amounts of H₂ after a 24-h digestion period

	Group	H ₂ : sludge (mL : mL)	CO ₂ (mL)	CH ₄ sludge (mL)	CH ₄ total (mL)	Inhibition (%)
Horizontal	A	0 : 42	8.69	6.56	6.56	0.00
		40 : 42	4.73	1.33	11.33	79.67
		60 : 42	2.96	0.72	15.72	89.04
		80 : 42	1.03	0.41	20.41	93.70
Vertical	G	40 : 42	5.16	4.12	14.12	37.22
		60 : 42	3.07	3.18	18.18	51.61
		80 : 42	1.20	3.00	23.00	54.24

Inhibition is the rate of inhibition of CH₄ production $(1 - \text{CH}_4\text{sludge} / 6.56) \times 100$ 6.56 (mL) is CH₄ production without H₂ addition

into one-unit volume of CH₄. The amount of CH₄ obtained from sludge digestion with H₂ can be calculated by quantifying the decrease in H₂ volume, because all of the H₂ is converted into CH₄ in accordance with Eqs. (5) and (6) :

$$\text{CH}_{4(a)} = \text{CH}_{4(\text{total, a})} - \text{CH}_{4(\text{H}_2 + \text{CO}_2, a)} \quad (5)$$

$$\text{CH}_{4(\text{H}_2 + \text{CO}_2, a)} = \text{H}_{2(a)} / 4 \quad (6)$$

CH_{4(a)} : volume of CH₄ produced from sludge while the initial volume of H₂ is a mL ;

CH_{4(total, a)} : total volume of CH₄ in sludge with added a mL H₂ ;

CH_{4(H₂+CO₂, a)} : volume of CH₄ arising from the conversion of a mL H₂ ;

H_{2(a)} : volume of H₂ ;

CH₄₍₀₎ : volume of CH₄ produced from sludge without added H₂.

The inhibition rate of digestion can be calculated as follows :

$$\text{Inhibition rate of digestion} = \text{CH}_{4(a)} / \text{CH}_{4(0)} \quad (7)$$

Based on Eq. (7), **Table 3** shows that digestion inhibition in Group G was significantly decreased. A significant increase in CH₄ production was observed (100 mL H₂ was not completely consumed; not included in the table). The rate of digestion inhibition decreased from 79.67%, 89.04% and 93.70% to 37.22%, 51.61% and 54.24% with the addition of 40, 60, and 80 mL H₂, respectively. On the basis of this result, H₂ should be injected from the middle or above instead of the bottom of single-phase digestion reactor or through a submerged membrane¹⁷⁾ to reduce sludge exposed to H₂.

4. Conclusion

In this study, the effects of exogenous H₂ on the thermophilic digestion of wastewater sludge from a Japanese treatment plant were investigated in batch experiments. The H₂ strongly inhibited the degradation of butyric acid and mildly inhibited the degradation of acetic acid. As an intermediate, the addition of H₂ inhibited both the acetogenesis process and syntrophic acetate oxidation.

Furthermore, the negative effect of H₂ could not be mitigated by decreasing the pH, adding CO₂, or decrease P_{H₂} but was decreased by reducing the size of the contact area between the sludge and H₂ (vertical setting of reactors). The digestion recovered

gradually when the P_{H₂} was very low. However, the recovery time was much longer than the H₂ consumption time. These phenomena disrupt the anaerobic digestion of wastewater sludge.

From an efficiency and economic point of view, our results suggest that the way of H₂ addition should follow the principle of reducing H₂ exposure such as adding from the middle or above instead of the bottom or through a submerged membrane in the single-phase digestion reactor. And given the high efficiency of CH₄ converted from H₂ in thermophilic digestion, ex-situ upgrading in an adjacent external reactor is a potential upgrading technology because it can be easily integrated with existing biogas plants while avoiding negative influences of exogenous H₂.

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高温消化下水汚泥への外部からの水素添加の影響

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概 要

本研究は、日本における高温消化下水汚泥に対し、外部から水素を添加した場合のバイオガス生成への影響をバッチ実験より調査した。対象汚泥では、主要なメタン生成古細菌は *Methanothermobacter* であったが、酢酸利用メタン生成菌としての *Methanosaeta* 属の存在量は、わずか 28% であった。

H_2 添加は酢酸生成と嫌気性酢酸酸化の両方を抑制し、その後メタン生成に影響を与えた。外因性水素の消化阻害による主な理由は、pH 増加や CO_2 減少ではなく、 H_2 濃度増加であった。水素分圧が非常に低くなると、消化は徐々に回復したが、回復時間は水素消費時間より非常に長くなった。

キーワード：水素、高温消化汚泥、メタン生成古細菌、酸生成細菌、嫌気性酢酸酸化細菌